Chromatography Equations

Plate Theory

p is fraction in mobile phase, q is fraction in stationary phase p + q = 1 $F_{(r, n)} = \frac{n!}{(n-r)!r!} p^{r} q^{n-r}$ Fraction $F_{(r, n)} = \frac{(np)^{r} e^{-np}}{r!} p^{r} q^{n-r} = \frac{(np)^{r} e^{-np}}{\sqrt{2\pi r}} \left(\frac{e}{r}\right)^{r}$ Fraction of solute in the rth of n plates. $r_{max} = np$ r_{max} is the peak, aka the plate with max solute fraction $F_{(r_{max},n)} = \frac{1}{\sqrt{2\pi r_{max}}}$ $Q_{(N,n)} = F_{(N,n)} m = \frac{m}{\sqrt{2\pi N}}$ Where $r_{max} = N$, then $Q_{(N,n)}$ is the quantity of material in the Nth plate $S_{max} = Q_{(N.n)} N/t_r$ $S_{max} = \frac{Nm}{\sqrt{2\pi N}} \frac{1}{t_r}$ $N = \frac{2\pi (S_{max})^2 t_r^2}{m^2}$ S_{max} is the rate of solute escape in mole/unit time Number of plates moles, k is proportionality constant, s is shape constant : 1/2 for triangular $m = s kht_w$ $S_{max} \propto h \text{ or } S_{max} = kh$ S_{max} (rate of solute escape) proportional to height of peak $\mathsf{N} = \frac{8\pi t_r^2}{t_w^2}$ N relates to column efficiency N= $16(\frac{t_R}{W_h})^2$ Efficiency (For gaussian peak shape) $N_{eff} = 5.54 (\frac{t'_R}{W_*})^2$ Effective efficiency - N using peak width at half height $A_s = BC/AC$ Peak Asymmetry - @10% peak h - AC is distance, from front to ctr, CB back to center $TF = AB/2AC = \frac{A_{5\%h} + B_{5\%h}}{2A_{5\%h}}$ Tailing Factor – measured at 5% h $\sigma = 0.3035 h$ h is peak height $\mathsf{N} = \left(\frac{t_r}{\sigma}\right)^2 = \frac{2\pi (S_{max})^2 t_R^2}{m^2}$ m is # moles HETP H = L/Nplate height is column length divided by # plates $H = \frac{L\sigma^2}{t_{\pi}^2} = A + \frac{B}{\mu} + C\mu$ $\mu_{opt} = \sqrt{\frac{B}{C}}$ Optimal mobile phase velocity $PC = 1 + \frac{\sqrt{N}}{4} \ln \left(\frac{t_R}{t_N}\right)$ Peak Capacity - # peaks can be fitted between t_M and t_R w/ 4σ gap H = A + B/u + CuVan Deemter equation, terms relate to Eddy diff, longit diff, $I = 100[n + (N - n)\left(\frac{\log(t_R)_u - \log(t_R)_n}{\log(t_R)_N - \log(t_R)_n}\right)]$ Kovats Index. Where I/100 = # of carbons

 \bar{v} = velocity of the retained solute u is velocity of the mobile phase $t_0 = t_M$ = Dead time or time of unretained solute

 $t_{r}' = t_{r} - t_{0}$ Adjusted retention time is retention time minus dead time $V_r' = V_r - V_0$ Adjusted Total volume $V_T = V_S + V_0$ $\bar{v} = \mu \frac{\text{moles of solute in mobile phase}}{\text{total number of moles}} = \mu \frac{c_M V_M}{c_M V_M + c_S V_S} = \mu \frac{1}{1 + \frac{c_S V_S}{c_M V_M}} = u \frac{1}{1 + \frac{K V_S}{V_M}} = \mu \frac{1}{1 + k'_A}$ $\bar{v}_A = \mu \; \tfrac{1}{1 + k'_A}$ velocity for retained solute A $k' = \left(\frac{u}{\bar{v}} - 1\right)^{A}$ Retention factor or capacity $k'_{A} = \frac{K_{A}V_{M}}{V_{M}} = \frac{t_{R}-t_{M}}{t_{M}} = \frac{t'_{R}}{t_{M}}$ $\kappa = \frac{V_{M}}{V_{S}} \left(\frac{u}{\bar{v}} - 1\right) = \frac{V_{M}}{V_{S}}k' = \beta k' = \frac{C_{S}}{C_{M}}$ Retention factor (capacity) for solute A Partition Factor / Distribution coefficient $\alpha = \frac{K_B}{K_A} = \frac{k'_B}{k'_A} = \frac{t'_R(B)}{t'_R(A)}$ $R_s = 2 \frac{t_{R2} - t_{R1}}{w_{b1} + w_{b2}}$ separation factor or selectivity Resolution (of peaks), w/ w_b being width at base $R_{s} = \frac{\sqrt{N}}{4} \frac{\alpha - 1}{\alpha} \frac{k}{k+1}$ $N = 16R_{s}^{2} \left(\frac{\alpha}{\alpha - 1}\right)^{2} \left(\frac{1 + k_{B}'}{k_{B}'}\right)^{2}$ Efficiency * Selectivity* Retention factor Efficiency from Rs, selectivity, and retention factor A_{UC} = 1.06 * h * $w_{0.5}$ Area under a gaussian curve $\beta = \frac{V_M}{V_S}$ $V_{\rm S} = L\pi r_c^2 L$ $K = \frac{V_R - V_0}{V_S}$ Vol of SP is $L\pi r_c^2 L$ of column, where r_c is radius of column $V_R = V_0 + KV_S$ $P' = P_1'C_1 + P_2'C_2$ P is polarity index, C is molar concentration $\frac{k_2'}{k_1'} = 10^{\frac{P_1' - P_2'}{2}}$ Changing Eluent polarity index affect, Normal phase $\frac{k_2'}{k_1'} = 10^{\frac{P_2'-P_1'}{2}}$ Changing Eluent polarity index affect, Reverse phase